

Year 7 Chemistry Curriculum – 2022-23

| | Autumn Term | | Spring Term | | Summer Term | |
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| | 1 | 2 | 1 | 2 | 1 | 2 |
| Key Concepts | Matter | | Reactions | | Earth | |
| National Curriculum Knowledge & Understanding | <p>Particle Model</p> <ul style="list-style-type: none"> * The properties of the different states of matter (solid liquid and gas) in terms of the particle model including gas pressure * Changes of state in terms of the particle model * Conservation of material and of mass and reversibility in melting freezing evaporation sublimation condensation dissolving * Similarities and differences including density differences between solids liquids and gases * Brownian motion in gases | <p>Pure and Impure substances</p> <ul style="list-style-type: none"> * The concept of a pure substance * Mixtures including dissolving and diffusion in terms of the particle model * Simple techniques for separating mixtures, filtration evaporation distillation and chromatography * The identification of pure substances | <p>Acids and Alkalis</p> <ul style="list-style-type: none"> * Naming acids and alkalis in terms of neutralisation reactions * The pH scale for measuring acidity/alkalinity and indicators * Chemical reactions as the rearrangement of atoms * Representing chemical reactions using equations | <p>Metals and Non-Metals</p> <ul style="list-style-type: none"> * The Periodic Table * The properties of metals and non-metals * Chemical reactions as the rearrangement of atoms * Representing chemical reactions using equations * The chemical properties of metal and non-metal oxides with respect to acidity * Reactions of acids with metals to produce a salt plus hydrogen * Reactions of acids with alkalis to produce a salt plus water | <p>Earth's Structure</p> <ul style="list-style-type: none"> * The composition of the Earth * The structure of the Earth * The rock cycle and the formation of igneous sedimentary and metamorphic rocks * Properties of ceramics | <p>Beyond the Atmosphere</p> <ul style="list-style-type: none"> * Our Sun as a star other stars in our galaxy other galaxies * The seasons and the Earth's tilt day length at different times of year in different hemispheres * The light year as a unit of astronomical distance * Use and derive simple equations and carry out appropriate calculations * Undertake basic data analysis including simple statistical techniques |

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| | * Diffusion in liquids and gases driven by differences in concentration * The difference between chemical and physical changes | | | | | |
| Assessment | Key Information 2 (Outdoor Key Information) arising from each unit of work covered separating mixtures | End of Term 0 Unit Assessment | Key Information 2 Teacher Assessment Acids and Alkalis | End of Term 0 Unit Assessment Reactions | Key Information 2 Teacher Assessment Earth | End of Year Assessment |
| Why this; Why now; | Basic understanding of the particle model is the heart of all branches of chemistry which is why it is the first module to be taught in year 9 chemistry. This must be studied at this point as the module covers pure substances in terms of the movement of particles. It references links | Pupils have prior knowledge of solutions from Key Information 1. This will be built on enabling pupils to explain how separation techniques work. Pure and Impure substances is taught early in year 9 as it develops aspects of the Particle Model module such as substances are made from particles and these can be a | Acids and Alkalis is taught at this point as pupils now know all substances are made up of particles. An understanding of neutralisation has to be addressed before pupils can begin to study the Metals and Non-Metals topic when different types of neutralisation reactions are investigated. | This learning module is at this point as it takes concepts from each of the previous modules: particles, solutions and neutralisation and sews them together while still opening opportunities for continual learning in subsequent modules. Examples in Types of | From Key Information 1 pupils will already know fossils are found in rocks. This unit is taught later in the year as pupils needed to previously learn about mixtures in Pure and Impure Substances and the concept of acids before chemical weathering in the rock cycle could be understood. This unit will explore | This topic links to Key Information 1 as pupils were taught about Earth and Space. This now triangulates making ties with the previous topic of Earth's Structure and the Particle Model as pupils will discover that the outer planets are made from gas whereas the inner planets are made from rock. Like the Earth. This unit builds a |

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